

res on crystalline barium
ts a similar behavior with

ree of pressurization and
%) intermediate pressures
0,000 psi and atmospheric
as low as 14,000 psi give
at atmospheric pressure.
0 psi represented in Figure
in the later stages of the
becomes increasingly sepa-
perature range 19–80°C.,
tive activation energy both
polymerizations, while the
d the crystalline dihydrate
°C. in the absence of pres-

ylate yield is increased by
mperature. The increased
essure and 100°C. may be
hange above 80°C.
e at 80 and 100°C. do not
esponding yields at room

n the course of the reaction
density differences for the
s might be expected to re-
ng fraction and magnitude
od of a propagating chain.
e polymer produced from
user than the monomer, the
le. If this is the case, then
ctorily explain our pressure
rylamide contradicts those
% volume reduction during
ecessary to repeat our work
determinations are carried
ll evaluation of our experi-
al analysis of the polymers
e and temperature, but the
ear in general terms to be

assistance of Mr. N. E. Sander-
al work.

References

1. A. J. Restaino, R. B. Mesrobian, D. S. Ballantine, and G. J. Dienes, *Rec. Sci.*, **25A**, 178 (1955).
2. A. J. Restaino, R. B. Mesrobian, D. S. Ballantine, G. J. Dienes, and D. J. Metz, *J. Am. Chem. Soc.*, **78**, 2939 (1956).
3. J. B. Lando and H. Morawetz, *J. Polymer Sci. C*, **4**, 789 (1964).
4. J. H. O'Donnell, B. McGarvey, and H. Morawetz, *J. Am. Chem. Soc.*, **86**, 2322 (1964).
5. G. Adler et al., *J. Polymer Sci.*, **44**, 117 (1960).
6. T. A. Fadner and H. Morawetz, *J. Polymer Sci.*, **45**, 475 (1960).
7. Y. Tabata, *Polymer Preprints*, **5**, 997 (1964).
8. C. H. Bamford, G. C. Eastwood, and J. C. Ward, *Nature*, **192**, 1036 (1961).
9. G. Adler and W. Reams, *J. Chem. Phys.*, **32**, 1698 (1960).